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Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.001 Å R factor = 0.036 wR factor = 0.101 Data-to-parameter ratio = 40.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-[N-(Prop-2-ynyl)-N-tosylamino]ethanol

Molecules of the title compound, $C_{12}H_{15}NO_3S$, are linked by a pair of $O-H\cdots O$ hydrogen bonds, forming a centrosymmetric $R_2^2(14)$ dimer. The dimers are cross-linked by a pair of $C-H\cdots O$ hydrogen bonds, forming a ribbon along the *a* axis.

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Comment

Benzenesulfonamide derivatives inhibit proliferation of colon adenocarcinoma cells (Kusunoki *et al.*, 2006) and are used as cyclooxygenase-2 (COX-2) inhibitors (Tsai *et al.*, 2006). They also exhibit antibacterial (Nieto *et al.*, 2005), anticancer and anti-HIV (Pomarnacka & Kozlarska-Kedra, 2003) activities. We report here the structure of the title compound, (I).



In the molecule of (I) (Fig. 1), the bond lengths and angles in the benzenesulfonamide unit (Table 1) agree with those reported for similar structures (Xing *et al.*, 2006; Zareef *et al.*, 2006). The sum of the bond angles around atom N1 (349.2°) indicates sp^3 hybridization.

Weak O-H···N and C-H···O interactions are observed in the molecular structure. In the solid state, the molecules are linked by a pair of O3-H3O···O1ⁱ hydrogen bonds, forming a centrosymmetric $R_2^2(14)$ dimer. The dimers translated by a unit cell along the *a* axis are linked by a pair of C10-H10···O3ⁱⁱ hydrogen bonds, forming a ribbon (Fig. 2); see Table 2 for symmetry codes.

Experimental

Propargyl bromide (1.2 mmol) was added dropwise to a solution of *N*-tosylaminoethanol (1 mmol) and anhydrous potassium carbonate



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(1.5 mmol) in dry acetone (20 ml). The mixture was stirred until the reaction was complete, as indicated by thin-layer chromatography. The reaction mixture was filtered. The filtrate was evaporated *in vacuo* and column chromatographed with a hexane–ethyl acetate (8:2) mixture to obtain the title compound. Crystals suitable for X-ray analysis were grown by slow evaporation of an ethyl acetate solution.

Z = 4

 $D_x = 1.353 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

 $\mu = 0.26~\mathrm{mm}^{-1}$

T = 100.0 (1) K

 $R_{\rm int} = 0.028$

 $\theta_{\rm max} = 37.5^\circ$

Block colourless

 $0.51 \times 0.44 \times 0.42 \text{ mm}$

26785 measured reflections

6524 independent reflections 5480 reflections with $I > 2\sigma(I)$

 $w = 1/[\sigma^2(F_0^2) + (0.0505P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

+ 0.279P]

 $(\Delta/\sigma)_{\rm max} = 0.001$

 $\Delta \rho_{\rm max} = 0.55 \text{ e } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.42 \text{ e} \text{ Å}^{-3}$

Crystal data

 $\begin{array}{l} C_{12}H_{15}NO_3S\\ M_r = 253.31\\ Monoclinic, P2_1/c\\ a = 8.3174 \ (2) \ A\\ b = 20.3109 \ (4) \ A\\ c = 7.4333 \ (1) \ A\\ \beta = 97.897 \ (1)^\circ\\ V = 1243.83 \ (4) \ A^3 \end{array}$

Data collection

Bruker SMART APEX2 CCD diffractometer ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 2005) $T_{\min} = 0.803, T_{\max} = 0.899$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.036$ $wR(F^2) = 0.101$ S = 1.05 6524 reflections 163 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1

Selected geometric parameters (Å, °).

S1-O2	1.4354 (6)	N1-C11	1.4749 (10)
S1-O1	1.4380 (6)	N1-C8	1.4800 (9)
S1-N1	1.6403 (7)	C8-C9	1.4687 (11)
S1-C1	1.7580 (8)	C9-C10	1.1962 (12)
O3-C12	1.4220 (10)		
O2-S1-O1	119.71 (4)	C11-N1-S1	115.87 (5)
C11-N1-C8	116.06 (6)	C8-N1-S1	117.22 (5)
C1-S1-N1-C11	-67.39 (6)	\$1-N1-C11-C12	-153.11 (5)
C1-S1-N1-C8	75.60 (6)	N1-C11-C12-O3	61.93 (8)

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
O3−H3O···N1	0.85(1)	2.62 (1)	2.9363 (9)	103 (1)
$O3-H3O\cdotsO1^{i}$	0.85 (1)	1.99 (1)	2.8227 (9)	166 (1)
C6-H6···O1	0.95	2.56	2.9258 (11)	103
$C8-H8A\cdots O1$	0.99	2.36	2.8732 (10)	111
$C10-H10\cdots O3^{ii}$	0.96(2)	2.38 (2)	3.2457 (12)	150 (1)
$C11-H11B\cdots O2$	0.99	2.44	2.9151 (10)	109

Symmetry codes: (i) -x, -y, -z; (ii) x + 1, y, z.

The propyne and hydroxyl H atoms were located in a difference map and refined isotropically. The C-bound H atoms were positioned



Figure 2

A view of the hydrogen-bonded (dashed lines) ribbon in (I). For the sake of clarity, H atoms not involved in the motif have been omitted.

geometrically and treated as riding on their parent C atoms, with C– H = 0.95–0.99 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ or $1.2U_{eq}(methyl C)$. A rotating-group model was used for the methyl group.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *APEX2*; data reduction: *SAINT* (Bruker, 2005); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1998); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

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